

1 **Quantification of polybrominated diphenyl ether (PBDE) congeners in**  
2 **wastewater by gas chromatography with electron capture detector (GC-**  
3 **ECD)**

4 **Oladapo Komolafe, Bernard Bowler, Jan Dolfing, Wojciech Mrozik and Russell**  
5 **Davenport \***

6 School of Engineering, Newcastle University, NE1 7RU, Newcastle Upon Tyne, United Kingdom

7 \* Corresponding author. Email: russell.davenport@ncl.ac.uk

8 **Abstract**

9 This paper describes a gas chromatography coupled with an electron capture detector (GC-  
10 ECD) method that was developed for screening and reliable quantification of some selected  
11 PBDE congeners (BDE 28, 47, 99, 100, 153, 154, 183 and 209) in wastewater. Emphasis was  
12 placed on the ability of the method to simultaneously analyse low to high BDE congeners in a  
13 single run, whilst being comparatively cost-effective to gas chromatography coupled with mass  
14 spectrometry (GC-MS) methods commonly employed today.

15 Different solid phase extraction (SPE) cartridges (Oasis HLB, Isolute PAH, and Isolute C18)  
16 were tested for efficient analyte extraction in terms of recovery and resultant clean extracts free  
17 of unwanted compounds. Isolute PAH performed better, and in combination with optimized  
18 GC-ECD conditions permitted satisfactory determination of PBDE congeners at trace levels  
19 (method detection limits (MDLs) of 0.6 ng/L to 11 ng/L) in water samples. Method accuracy  
20 and precision were evaluated by recovery experiments using laboratory spiked water samples  
21 at two concentrations (3 ng/L and 10 ng/L).

22 The method was employed to evaluate the fate and removal of selected PBDE congeners in a  
23 conventional activated sludge wastewater treatment plant (WWTP) in Northern England.  
24 PBDEs were detected in influent and effluent samples at 0 – 113 ng/L and 0 – 18 ng/L  
25 respectively, and 75% total PBDE removal was achieved by the WWTP.

26

27 **Keywords:** Polybrominated diphenyl ethers; Solid phase extraction; Gas chromatography;  
28 Wastewater analysis

## 29 **1. Introduction**

30 Brominated flame retardants (BFRs) are added to many household items such as furniture,  
31 upholstery, plastic, electronic devices and textiles; they consist of several groups of compounds,  
32 including polybrominated diphenyl ethers (PBDEs), hexabromocyclododecane (HBCDD) and  
33 tetrabromobisphenol A (TBBPA)<sup>1-3</sup>. PBDEs are the second most produced BFR in the world,  
34 accounting for 33% of global production, while TBBPA accounts for 60%<sup>4</sup>. According to the  
35 United Nations Environment Programme, the total global production of all PBDEs from 1970  
36 to 2005 was between 1.3 million and 1.5 million tonnes with deca-mix BDE formulation  
37 accounting for about 85 % of this number<sup>5</sup>. PBDEs are persistent and hydrophobic compounds  
38 that tend to bio-accumulate; their occurrence in the environment has been of growing concern  
39 due to their toxicological effects including disruption of the thyroid hormone function in  
40 humans and wildlife, which has led to a ban on their production and usage in the European  
41 Union (EU) and the United States of America (USA)<sup>4,6</sup>. In the EU, production, import and use  
42 of commercial PBDE formulations, including pentaBDE and OctaBDE, was banned in 2003<sup>7</sup>,  
43<sup>8</sup>, and this was extended to a worldwide ban in 2009 at the Stockholm Convention of Persistent  
44 Organic Pollutants<sup>9</sup>. Despite this ban, these PBDEs are still being deposited into the  
45 environment, as they are not covalently bound to the applied products and are released with  
46 usage<sup>10,11</sup>. Deca-BDE formulation is currently being phased out in the EU and US, but still  
47 extensively produced in China and the resulting products are distributed globally<sup>12</sup>.

48 Effluents from WWTPs have been identified as a major source of PBDE into the environment.  
49 WWTPs receive PBDEs through municipal wastewater (discharged during production,  
50 application and release from in-use domestic products) and surface runoff<sup>10,12</sup>. Therefore, an  
51 effective analytical methodology to extract and quantify the presence of PDBEs in wastewater  
52 is required.

53 Several methods, including conventional liquid-liquid extraction (LLE), pressurized liquid  
54 extraction (PLE), solid phase extraction (SPE), solid phase micro-extraction (SPME)  
55 ultrasound assisted extraction and cloud point extraction (CPE) have been used to extract  
56 PBDEs from water and semi-solids <sup>4, 13, 14</sup>. Most of these methods have limitations, including  
57 excessive use of solvents, and/or require further sample clean up using multilayer column  
58 chromatography or gel permeation chromatography to reach lower detection limits <sup>14, 15</sup>. SPE  
59 is the most common extraction method used today as it offers low solvent usage and broad  
60 applicability. Careful selection of appropriate sorbent and an optimized elution protocol can  
61 produce clean extracts that require no further clean-up <sup>16, 17</sup>. For unequivocal identification and  
62 quantification of brominated flame-retardants including PBDEs in environmental samples, gas  
63 chromatography coupled with mass spectrophotometry (GC-MS) operating in either electron  
64 ionization (EI) or electron capture negative ionization (ECNI) mode is often used <sup>13, 18, 19</sup>.  
65 However, GC-ECD (electron capture detector) has also been employed for quantification of  
66 PBDEs in environmental samples <sup>15, 20, 21</sup>.

67 Determination of PBDEs in environmental samples is most commonly carried out by GC-MS  
68 mainly because of its selectivity, but GC-ECD is advantageous as it is cheaper, more user  
69 friendly and more sensitive, due to lower detection limits <sup>22</sup>. However, the ECD is prone to  
70 halogenated interference, since identification and resulting quantification of compounds is  
71 solely based on retention time; hence ECD based methods suffer from limited selectivity <sup>22</sup>.  
72 This shortcoming of the GC-ECD method can, however, be effectively minimised by carefully  
73 selecting GC columns and a clean-up method that produces high quality extracts free of  
74 interfering compounds. GC-ECD system is cheaper than GC-MS systems for micropollutant  
75 analysis in terms of purchasing, maintenance and costs per sample. For example, a basic model  
76 GC-ECD system by Agilent Technologies costs half the price of a GC-MS system in both  
77 capital (about \$45,000 for GC-ECD and \$91,000 for GC-MS) and maintenance cost (\$1200 for

78 GC-ECD and \$2600 for GC-MS per maintenance visit). Additionally, private laboratories  
79 carrying out analysis of total polychlorinated biphenyls (PCB) and congeners in soil and  
80 sediments reported that per sample GC-MS costs (\$500 - \$1000) are twice those of GC-ECD  
81 (\$250 - \$750) <sup>23</sup>. This method will be even more useful in low-middle income countries  
82 (LMICs) where shortage of studies on chemical pollution has been attributed to difficulties and  
83 high costs associated with environmental analysis <sup>24</sup>.

84 To the best of our knowledge, only a few researchers have applied GC-ECD to analyse PBDEs  
85 in water and wastewater <sup>15, 20</sup>, and none of them has used SPE for sample concentration and  
86 extraction. The authors from previous studies employed LLE, CPE or molecularly imprinted  
87 SPME and were unable to analyse BDE 209 (which is a major component of the decaBDE mix  
88 mentioned above) together with other lower molecular weight congeners in one single run: a  
89 separate GC capillary column and temperature program was generally required. In this work,  
90 a cost-effective yet functional analytical technique was developed using the combination of  
91 SPE and GC-ECD for the determination of selected low to high molecular weight PBDE  
92 congeners (BDE 28, BDE 47, BDE 99, BDE 100, BDE 153, BDE 154, and BDE 209) (Table  
93 *I*) in wastewater; special attention was given to the determination of BDE-28 to BDE 209 in a  
94 single run. The method, which involved using a multilayer SPE column for extraction and  
95 analytical performance, was evaluated on the basis of its detection limits, linear working range  
96 and repeatability. The method was then applied to determine the levels and removal of PBDEs  
97 in a municipal wastewater treatment plant in Northern England.

## 98 **2. Materials and methods**

### 99 *2.1 Materials and reagents*

100 A certified standard solution mix of PBDEs (> 98% purity) containing eight primary congeners  
101 including BDE 28, BDE 47, BDE 99, BDE 100, BDE 153, BDE 154, BDE 183 and BDE 209  
102 was obtained from Accustandard Inc via Kinesis (UK). The concentration of the congeners was  
103 2.5 µg/ml in isooctane, except BDE 209 which was present at 25 µg/ml. BDE 77 (50 µg/ml in  
104 isooctane), PCB 209 (10 µg/ml in heptane) and 4PC-BDE-208 (50 µg/ml in toluene), which  
105 were used as surrogate standards, were purchased from Accustandard (via Kinesis UK), Sigma  
106 Aldrich (UK), and Wellington Laboratories (via Greyhound Chromatography UK) respectively,  
107 with purities higher than 98%.

108 Stock solutions were prepared by dissolving the reference and surrogate standard in acetone  
109 (at 500 ng/ml). Working solutions were then prepared by diluting the stock solutions in acetone  
110 for sample fortification and in ethyl acetate for instrumental analysis. All solutions were stored  
111 at 4 °C, and were allowed to reach room temperature for 15 minutes before use. Ultra-trace  
112 grade of acetone, ethyl acetate and isopropanol were obtained from Sigma Aldrich (UK). Oasis  
113 HLB cartridges (200mg, 6cc) were purchased from Waters (UK); Isolute C<sub>18</sub> (1g, 6 ml), ABN  
114 and Isolute PAH cartridges (1.5g, 6 ml) were from Biotage (UK).

### 115 *2.2 Sample collection*

116 Grab wastewater samples (raw influent and final effluent) were collected from a nitrifying  
117 activated sludge treatment plant in North East England with a population equivalent of 22,500.  
118 Samples were collected in cleaned and disinfected (with 1% Virkron for 24 hours, then rinsed  
119 multiple times with distilled water to get rid of chlorine residues) HDPE containers - analysis  
120 of containers did not show any contamination with target compounds. Samples were stored at

121 4 °C upon arrival to the laboratory and were used within 24 hours. To account for the  
122 concentration of the chemicals in both aqueous and particulate matter, half of the samples were  
123 filtered through glass microfiber filter to estimate the concentration of the chemicals in the  
124 aqueous phase alone (Sartorius MGB filters, 0.7mm thick, 1.0 µm particle retention) before  
125 being passed through the SPE cartridges. The other half of the samples was not filtered before  
126 SPE.

### 127 *2.3 GC Instrumentation*

128 Analyses were performed on an Agilent 7890A gas chromatography system equipped with  
129 micro electron capture detector (µECD) and an Agilent 7683B automatic injector. A DB-5MS  
130 fused silica capillary column (15m x 0.25mm I.D. x 0.1µm film thickness) from J & W  
131 Scientific, USA was used for chromatographic separation. The inlet was fitted with an SGE  
132 single taper deactivated glass liner and samples were injected (1 µl) in split/splitless mode (1  
133 min. splitless, then 30ml/min split). The inlet and detector temperature was set at 300 °C and  
134 290 °C respectively. H<sub>2</sub> (99.999%) was used as carrier gas in constant flow mode (flow-rate 1  
135 ml/min, velocity 91.3 cm sec<sup>-1</sup>). The µECD was used with N<sub>2</sub> (99.999%) make-up gas at a  
136 flow-rate of 30 ml/min. The GC oven was heated from 100 °C (initial hold time 1 min) to  
137 150 °C (hold time 0 min) at 50 °C/min. and then to 290 °C (final hold time 15 min) at 12.5  
138 °C/min. Data was acquired and processed using a Thermo Atlas chromatographic data system  
139 (version. 8.3). Analytes were identified solely by their retention time and quantified by their  
140 integrated peak area.

141 The retention time of all eight PBDE congeners was confirmed using an Agilent 7890A GC  
142 split/split less injector linked to an Agilent 5975C MSD to ensure correct identification of  
143 analyte. This GC-MS was equipped with the same DB-5MS capillary column (15m x 0.25mm  
144 I.D. x 0.1µm film thickness) used above. Analysis was done in EI mode, full scan and SIM

145 spectrum was acquired. Ions (m/z) corresponding to the fragmentation of each BDE congeners  
146 was monitored to add a level of confidence in the identification of the analytes (Table 2).

#### 147 *2.4 SPE procedure and optimization*

148 To achieve the low detection limits required for the quantification of analytes at environmental  
149 levels, the performance of SPE cartridges including Oasis HLB (200mg, 6ml), Isolute C18 (1g,  
150 6ml) and Isolute PAH (1.5g, 6ml) was evaluated for the extraction of PBDEs from wastewater  
151 samples. Oasis HLB and C18 cartridges are popular choices for the extraction of PBDEs from  
152 environmental samples and clean-up<sup>16, 17, 25, 26</sup>. The extraction procedure was modified from a  
153 Biotage application note for analysing polycyclic aromatic hydrocarbons (PAHs) in water<sup>27</sup>.  
154 Briefly, isopropanol (1% v/v) was added to 100 ml of filtered wastewater sample without pH  
155 adjustment for extraction of PBDEs in the aqueous phase, followed by addition of surrogate  
156 standards at 5 ng/L each for BDE 77, PCB 209 and 50 ng/L for 4PC-BDE-208, before passing  
157 through the SPE. Cartridges were conditioned with 5 ml of isopropanol followed by deionized  
158 water containing 2 % isopropanol (v/v) at a flow rate of 5 ml/min. Samples were then passed  
159 through the cartridges at a flow rate of 10 ml/min. Sample bottles were then rinsed with acetone  
160 to prevent loss of analyte to the glass walls, and then diluted with 90 ml of deionized water  
161 before passing through the cartridges. This also serves as a prewash step to remove unwanted  
162 impurities including chlorophyll from the cartridges. The cartridge was finally washed with  
163 10 % isopropanol (v/v) and dried under vacuum for 45 minutes. Elution was performed by  
164 passing 5 ml of ethyl acetate through the cartridge twice (10 ml total volume). The extract was  
165 then evaporated under a gentle stream of nitrogen at 30 °C using Labconco Rapidvap  
166 Evaporator to 500 µl before injection into the GC-ECD. Furthermore, between 50 ml – 200 ml  
167 of spiked water samples were tested, and 100 ml sample load performed better in terms of  
168 adequate analyte extraction/retention and low background noise. Different volumes of ethyl  
169 acetate were also tested to optimize analyte elution, and 10 ml (5 ml twice) showed optimum



170 result and was employed in further work. For extraction of PBDEs in both the aqueous and  
171 particulate matter (PM) phase, 20 ml and 50 ml of unfiltered influent and effluent was diluted  
172 to 100 ml with deionized water respectively, and was processed as the filtered samples above.

### 173 *2.5 Method validation study*

174 Analytes were identified mainly by their retention time and the calibration curves were  
175 generated by injecting reference PBDE standard in triplicate at five concentration levels: from  
176 0.5 to 10 ng/ml for all PBDEs except BDE 209 which ranged from 5 – 100 ng/ml, because of  
177 its lower sensitivity to the ECD detector and relatively higher environmental concentrations.  
178 Linearity was observed when the correlation coefficient was  $> 0.99$ . To accurately predict the  
179 concentration of an unknown analyte in a sample- especially at trace levels, an appropriate  
180 calibration model is important. Selecting an appropriate calibration model involves a decision  
181 as to either allow the calibration curve pass through a point of intercept on the y-axis (so that  
182  $y = mx + c$ ) or force the curve to go through the origin (so that  $y = mx$ )<sup>28</sup>. Regression statistics  
183 of the calibration data on Microsoft Excel 2013 can be used to make this decision. This decision  
184 is based on closeness of the y-intercept to zero, and can be tested statistically using standard  
185 error (SE)<sup>28</sup>. The standard error of the y-intercept ( $SE_y$ ) obtained by the regression analysis is  
186 based upon the variability at the y-intercept and can be used to test if the curve passes through  
187 zero such that;

188 When y-intercept  $> SE_y$ , use intercept such that  $y = mx + c$

189 When y-intercept  $\leq SE_y$ , force curve through the origin, such that  $y = mx$  ( $c = 0$ ).

190 The determined calibration curve model and equation for all eight BDE congeners is given in  
191 Table 3.

192 PCBs have similar physical-chemical properties to BDEs <sup>29</sup>, therefore rare PCB congeners can  
193 be used as surrogates for PBDE determination. BDE 77 and PCB 209 were primarily selected  
194 for internal standard quantification for BDE 28 to BDE 183 because of their absence from  
195 wastewater; these compounds have been used for this purpose in previous studies <sup>15, 17, 25</sup>.  
196 However, BDE 77 was dismissed as a surrogate when initial analysis of environmental samples  
197 showed its presence in wastewater. 4PC-BDE-208 was employed as surrogate standard for  
198 BDE 209. This compound was proposed by Wellington Laboratories (Canada) as the perfect  
199 surrogate for decaBDEs when an instrumental method without the capability to differentiate  
200 between mass-labelled and parent compound is employed, such as an GC-ECD method <sup>30</sup>.  
201 Additionally, the similarity in structure/chemical composition to BDEs, and absence in  
202 environmental samples supports its suitability as a surrogate standard. The effect of complex  
203 sample matrix on GC-ECD analysis was investigated by spiking the PBDE congeners at 10  
204 ng/L into deionized water (DI), influent and effluent (n = 3). The spiking was performed after  
205 sample extraction with SPE and prior to GC sample injection. DI water was used as  
206 blank/absolute recovery and this recovery was compared to recoveries recorded in the influent  
207 and effluent samples.

208 Method accuracy was evaluated by performing recovery experiments in blanks (deionized  
209 water, n = 3) and matrix samples (final effluent, n = 3) at two fortification levels (3 ng/L and  
210 10 ng/L, 30 ng/L and 100 ng/L for BDE 209). The repeatability of the method was determined  
211 by the relative standard deviation (% RSD) from the recovery experiments in the fortified blank  
212 and matrix sample <sup>31, 32</sup>. Instrumental limits of detection (IDL) were established as the lowest  
213 analyte concentration that gave a signal to noise ratio of three (s/n = 3) upon the injection of  
214 standard solutions, and was determined on the Atlas software. The method detection limit  
215 (MDL) was determined according to EPA method 1984 <sup>33</sup>. Briefly, analytes were spiked at a  
216 concentration of between one to five times of IDL in blank (DI water, n = 7) and matrix sample

217 (effluent,  $n = 7$ ), then analysed on the GC-ECD. The resultant standard deviation was  
218 multiplied by the students' T-value that corresponds to six degrees of freedom to estimate the  
219 MDL.

$$220 \quad MDL = T_{(n-1, 1-\alpha=0.99)} * (S)$$

221 Where:  $T_{(n-1, 1-\alpha=0.99)}$  = students' T value for a 99% confidence level, and a standard  
222 deviation estimate with  $n - 1$  degrees of freedom.

223 S = standard deviation of replicate analyses.

224 The method quantification limit (MQL) was set at three times the MDL.

## 225 **3. Results and discussion**

### 226 *3.1 Chromatographic performance*

227 Splitless injection mode is preferred for PBDE analysis due to the relatively low environmental  
228 levels of these compounds, and a high injection temperature (300 °C and above) is  
229 recommended to minimise discrimination of high molecular weight congeners and thermal  
230 degradation<sup>22</sup>. Chromatographic separation optimization is necessary when analysing a wide  
231 range of BDE congeners (from triBDEs to decaBDEs) as this ensures good separation of  
232 compounds of interest when using a one-column approach. Employing a short column (15 m)  
233 with a short internal diameter ( $\leq 0.25$  mm) allowed the detection and quantification all eight  
234 BDE congeners including BDE 209 that easily degrades in GC column when subjected to high  
235 temperatures, without compromising separation, as reported by other authors<sup>8, 14</sup>. Using a  
236 relatively high carrier gas flow rate further helped to reduce the degradation of BDE 209 within  
237 the GC inlet and column. This optimization made it possible to use one GC column instead of  
238 two separate GC columns: one for lighter BDEs and one for BDE 209, as used in previous  
239 studies for GC-ECD analysis<sup>15, 20</sup>. The optimized temperature program in addition to the  
240 capillary column applied allowed for the separation of the BDE congeners in under 16.5 min,  
241 with BDE 28 eluting first at 5.3 min and BDE 209 at 16.4 mins (Figure 1; Figure 2).

### 242 *3.2 SPE optimization and method performance*

243 The extraction procedure was optimized by testing different SPE cartridges for their efficiency  
244 in analyte retention. Cartridges tested included Oasis HLB (200 mg), HLB prime, C18 (1 g),  
245 and Isolute PAH (1.5 g). Oasis HLB and C18 have been popular choices in the literature for  
246 extraction of BDEs from water samples<sup>16, 17, 25</sup>. The result of preliminary recovery experiments  
247 of eight PBDE congeners spiked in water samples showed that Isolute PAH performed better  
248 than the other cartridges, especially in filtered effluent samples. With Isolute PAH recovery of

249 analytes was between 69 – 126 % with a relative standard deviation (RSD) of less than 13%  
250 (Table 4), and mean recoveries in effluent were significantly greater than with the Oasis HLB  
251 cartridge (Mann-Whitney test,  $P < 0.05$ ) and slightly but not statistically significantly greater  
252 than with Isolute C18 (Mann-Whitney Test,  $P = 0.06$ ). The recoveries were not significantly  
253 different in DI water. Extracts from the Isolute PAH cartridge showed the least interference  
254 and lower background noise. The cartridge comprises an octadecyl layer (1000 mg) with an  
255 amino based sorbent (500 mg) that helps remove polar interferences such as humic acids from  
256 the effluent according to Biotage (the cartridge manufacturer), thereby reducing background  
257 noise and improving analyte recovery. Therefore, the Isolute PAH was selected for further  
258 optimization and subsequent analysis.

259 The matrix effect test showed that peak signals of the PBDE congeners were enhanced in  
260 influent and effluent samples, except for BDE 99, which was suppressed in effluent (Figure 3).  
261 The matrix effect was about -31% (BDE 99) to 91% (BDE 209), and 3.2% (BDE 99) to 65%  
262 (BDE 209) in influent and effluent samples respectively (Figure 3). The matrix effect also  
263 appears to be generally higher in influent than in effluent, which might be expected due to the  
264 presence of more organic matter in the influent. This observed signal modulation implies that  
265 surrogate standards are needed to accurately quantify PBDE congeners in wastewater.

266 Further recovery studies were carried out on effluents with the Isolute PAH cartridge at two  
267 fortification levels; 3 ng/L and 10 ng/L for all BDEs, except BDE 209, which was studied at  
268 30 ng/L and 100 ng/L because of its relatively higher concentrations in wastewater. Recoveries  
269 were surrogate standard-corrected using PCB 209 and 4PC-BDE-208 as standards. The US  
270 EPA method 1614 recommends analyte recovery of 60 – 140 % for BDE 28 – 183, and 50 –  
271 200% for BDE 209, with an RSD less than 40 % for initial demonstration of method precision  
272 and accuracy<sup>34</sup>. At 3 ng/L fortification level (Table 5) good recoveries (78- 135%) were

273 recorded for all analytes except BDE 47 (176%). Recoveries of PBDEs at 10 ng/L were mostly  
274 between the acceptable ranges of 60 – 129 %, except BDE 183 with recovery of 150 % (**Error!**  
275 **Reference source not found.**). RSDs were generally lower than 20% thereby showing method  
276 precision.

277 As shown in Table 4, the MDL of the eight PBDE congeners in water ranged from 0.14 ng/L  
278 to 10 ng/L in deionized water and 0.2 ng/L to 10.8 ng/L in effluent samples. These values are  
279 below the reported levels of these chemicals in wastewater around the world <sup>10, 25, 35</sup>, hence the  
280 method can be used globally to quantify them accurately. The method detection limits achieved  
281 were also well below the maximum allowable concentration environmental quality standards  
282 (MAC-EQs; 140 ng/L and 14 ng/L for inland and other surface waters respectively) for  
283 PBDEs (sum of congeners BDEs 28, 47, 99, 100, 153 and 154) as proposed in the EU Water  
284 Framework Directive (WFD) <sup>36</sup>.

### 285 *3.3 Application to wastewater samples*

286 The developed method was employed to quantify the concentrations of eight primary PBDE  
287 congeners in both the dissolved and particulate matter phase of municipal wastewater, and  
288 assess the mass removal rate achieved by the treatment plant. The PBDE congener profile in  
289 influent and effluent showed the presence of seven congeners; BDE 100 was not detected  
290 (Figure 4). The total (aqueous and particulate matter) concentration of the individual PBDEs  
291 was between 2.1±0.3 – 111±10.7 ng/L ( $\Sigma$ PBDE = 169 ng/L) in raw influent and 1.6±0.3 –  
292 17.7±1.4 ng/L ( $\Sigma$ PBDE = 43 ng/L,  $\Sigma$ PBDE EU WFD congeners = 19 ng/L) in the final effluent  
293 (Table 6). These concentrations are an order of magnitude lower than the EU WFD MAC-EQS  
294 and are similar to reported levels in Canada <sup>10</sup> Australia <sup>37</sup>, and China <sup>25</sup>. Risk assessment for  
295 BDE 209 was performed as reported by Cristale *et al.*, 2013 <sup>16</sup> since there is no specific EQS  
296 value for this single congener. The predicted no effect concentration (PNEC) for aquatic

297 organisms (including fishes, daphnids and algae) was calculated as 4.8 ng/L and maximum  
298 measured concentration (MC) was used to obtain a risk quotient ( $RQ = MC/PNEC$ ). Low to  
299 significant adverse effect of BDE 209 was indicated when  $1.0 \leq RQ < 10$ , and  $10 \leq RQ < 100$   
300 respectively. In the present study, low potential for adverse effects on aquatic organisms was  
301 observed for BDE 209 in the effluent of the WWTP, with an RQ of 3.7.

302 The concentration of BDE 209 represented 66% of the total PBDE concentration in influent  
303 and 40% in effluent samples (Figure 4). BDE 209 has been reported to dominate the total  
304 concentration of PBDE found in wastewater around the world<sup>10, 35, 38</sup>. Furthermore, only BDE  
305 209 was detected in a river receiving effluent from wastewater treatment plants in the United  
306 Kingdom, to a concentration of up to 290 ng/L which were associated with significant levels  
307 of risk to aquatic wildlife<sup>16</sup>. About 82% and 49% of the total PBDE was present in the  
308 particulate phase of the influent and effluent respectively (Figure 5). This is as expected due to  
309 the high log  $K_{OC}$  values (Table 1) of PBDEs, which gives rise to a high association with  
310 suspended solids. This observation is in line with previous findings that over 90% of PBDEs  
311 in influent tends to absorb to sludge in WWTP<sup>39, 40</sup>. About 75% removal of total PBDE  
312 concentration was achieved by the WWTP; this removal is most likely due to partitioning and  
313 settling out with the sludge in the primary and secondary sedimentary tanks<sup>39</sup>. PBDE  
314 concentrations in sludge are reported to vary over one order of magnitude<sup>19, 41</sup>, while  
315 concentrations in anaerobically treated sludge are unknown. The potential risks PBDEs pose  
316 via sludge applications to soil are therefore also unknown. Furthermore, the distribution of  
317 PBDE over the dissolved and particulate matter (PM) phases in influent and effluent samples  
318 (Figure 5) indicates that PBDEs were mostly removed via the PM phase in the WWTP. It must  
319 be noted that wastewater treatment is the preserve of high-income countries and cities, with  
320 80% of the world's wastewater going untreated into receiving watercourses<sup>42</sup>. Concentrations

321 of PBDEs in domestic wastewater may be at levels that cause a risk to aquatic wildlife in highly  
322 populated urban centres, especially in low to middle income countries.



## 323 4. Conclusions

324 In this study, a novel SPE-GC-ECD method was developed for the analysis of eight primary  
325 PBDE congeners in wastewater. The extraction procedure was optimized by testing different  
326 SPE cartridges including Oasis HLB, Isolute C18 and Isolute PAH. The Isolute PAH cartridge  
327 proved to be superior in sample clean up and extracting low and high molecular weight PBDEs  
328 in wastewater, perhaps due to its unique combination of a C18 and amino based sorbent. The  
329 detection of the PBDEs by this method was corroborated using GC-MS. The low detection  
330 limits obtained allowed for determination of PBDEs at environmentally relevant levels, and,  
331 importantly, were well below the proposed MAC-EQs for PBDEs set by the EU Water  
332 Framework Directive. Although the method presented here may have limitations due to a lack  
333 of selectivity of the GC-ECD compared to a GC-MS/MS system, it is a more cost-effective  
334 solution for quantifying PBDE concentrations in wastewater, especially if widespread  
335 monitoring is required. GC-ECD is two-fold cheaper than GC-MS in capital costs and up to  
336 four-fold cheaper in operational costs since all PBDE congeners can be analysed in a single  
337 run. This method thereby opens up PBDE analysis to laboratories without a GC-MS system or  
338 can allow laboratories to commit GC-MS systems to other functions. This will be especially  
339 useful in developing countries with limited resources to carry out environmental analyses and  
340 could form the basis of a traffic light system whereby water bodies potentially at risk could  
341 first be flagged using this method for subsequent confirmation using the more expensive GC-  
342 MS method. If we rely solely on the best most expensive methods, we would miss the broader  
343 picture of a global chemical pollution; which of-course starts with the ability to detect and  
344 measure these chemicals in water.

345 The GC-ECD method was employed for the quantification of PBDEs at trace levels in the  
346 influent and effluent of a UK conventional activated sludge WWTP. All BDEs of interest were

347 detected in both influent and effluent samples at trace concentrations, that is, in the 2-20 ng/L  
348 range, except BDE 209, which was present at about tenfold higher concentrations, and BDE  
349 100, which was not detected. BDE 209 was the most abundant analyte, and 50 – 80% of the  
350 BDEs partitioned onto the particulate matter. Finally, about 75% removal of the total PBDE  
351 concentration was achieved by the WWTP, resulting in levels assessed to pose little but not  
352 negligible risk to aquatic wildlife. PBDE removal was most likely due to adsorption onto sludge  
353 during secondary treatment. Urban areas with little or no wastewater treatment might not be  
354 afforded a similar level of environmental protection against PBDEs as offered by the WWTP  
355 in our study area.

356

### 357 **Conflict of interest**

358 The authors have no potential conflicts of interest.

359

360 **Acknowledgements**

361 This project was funded by a Challenging Engineering award from the Engineering and  
362 Physical Science Research Council (EPSRC; EP/I025782/1).

## 363 References

- 364 1. Z. Xie, R. Ebinghaus, R. Lohmann, O. Heemken, A. Caba and W. Püttmann, *Analytica chimica*  
365 *acta*, 2007, **584**, 333-342.
- 366 2. USEPA, Technical Fact Sheet – Polybrominated Diphenyl Ethers (PBDEs) and Polybrominated  
367 Biphenyls (PBBs), 2014,
- 368 3. F. Rahman, K. H. Langford, M. D. Scrimshaw and J. N. Lester, *Science of the Total*  
369 *Environment*, 2001, **275**, 1-17.
- 370 4. P. Labadie, K. Tlili, F. Alliot, C. Bourges, A. Desportes and M. Chevreuril, *Analytical and*  
371 *bioanalytical chemistry*, 2010, **396**, 865-875.
- 372 5. UNEP, *Guidance for the inventory of polybrominated diphenyl ethers (PBDEs) listed under the*  
373 *Stockholm Convention on Persistent Organic Pollutants*, Geneva, 2015.
- 374 6. M. Gorga, E. Martinez, A. Ginebreda, E. Eljarrat and D. Barcelo, *Science of the Total*  
375 *Environment*, 2013, **444**, 51-59.
- 376 7. J. Cristale, J. Quintana, R. Chaler, F. Ventura and S. Lacorte, *Journal of Chromatography A*,  
377 2012, **1241**, 1-12.
- 378 8. O. Hutzinger, B. Beek and M. Metzler, *The handbook of environmental chemistry*, Springer,  
379 2013.
- 380 9. EU, Stockholm Convention on Persistent Organic Pollutants; Recommendations of the  
381 persistent organic pollutants review committee of the Stockholm convention to amend  
382 annexes A, B or C of the convention, 2009, (accessed 15 August 2017).
- 383 10. M. Kim, P. Guerra, M. Theocharides, K. Barclay, S. A. Smyth and M. Alaei, *Water research*,  
384 2013, **47**, 2213-2221.
- 385 11. M. A. Siddiqi, R. H. Laessig and K. D. Reed, *Clinical Medicine & Research*, 2003, **1**, 281-290.
- 386 12. N. Xiang, L. Chen, X.-Z. Meng, Y.-L. Li, Z. Liu, B. Wu, L. Dai and X. Dai, *Science of the Total*  
387 *Environment*, 2014, **487**, 342-349.
- 388 13. A. R. Fontana, M. F. Silva, L. D. Martínez, R. G. Wuilloud and J. C. Altamirano, *Journal of*  
389 *Chromatography A*, 2009, **1216**, 4339-4346.
- 390 14. A. Covaci, A. C. Dirtu, S. Voorspoels, L. Roosens and P. Lepom, in *Brominated Flame*  
391 *Retardants*, Springer, 2010, pp. 55-94.
- 392 15. A. P. Daso, O. S. Fatoki, J. P. Odendaal and O. O. Olujimi, *Archives of environmental*  
393 *contamination and toxicology*, 2012, **62**, 391-402.
- 394 16. J. Cristale, A. Katsoyiannis, A. J. Sweetman, K. C. Jones and S. Lacorte, *Environmental*  
395 *pollution*, 2013, **179**, 194-200.
- 396 17. J. Sánchez-Avila, J. Bonet, G. Velasco and S. Lacorte, *Science of the Total Environment*, 2009,  
397 **407**, 4157-4167.

- 398 18. B. Mai, S. Chen, S. Chen, X. Luo, L. Chen, L. Chen, Q. Yang, G. Sheng, P. Peng and J. Fu,  
399 *Environmental Science & Technology*, 2005, **39**, 3521-3527.
- 400 19. S. Lee, G. J. Song, K. Kannan and H. B. Moon, *Sci Total Environ*, 2014, **470-471**, 1422-1429.
- 401 20. M. K.-Y. Li, N.-Y. Lei, C. Gong, Y. Yu, K.-H. Lam, M. H.-W. Lam, H. Yu and P. K.-S. Lam,  
402 *Analytica chimica acta*, 2009, **633**, 197-203.
- 403 21. C. A. De Wit, *Chemosphere*, 2002, **46**, 583-624.
- 404 22. H. M. Stapleton, *Analytical and bioanalytical chemistry*, 2006, **386**, 807-817.
- 405 23. F. T. Price, K. V. Brix and N. K. Lane, *Environmental Toxicology and Risk Assessment: Recent  
406 Achievements in Environmental Fate and Transport*, Vol. 9, ASTM, 2000.
- 407 24. G. P. Pessoa, N. C. de Souza, C. B. Vidal, J. A. Alves, P. I. Firmino, R. F. Nascimento and A. B.  
408 Dos Santos, *Sci Total Environ*, 2014, **490C**, 288-295.
- 409 25. D. Deng, H. Chen and N. F. Y. Tam, *Science of the Total Environment*, 2015, **502**, 133-142.
- 410 26. M. Gorga, E. Martínez, A. Ginebreda, E. Eljarrat and D. Barceló, *Science of the Total  
411 Environment*, 2013, **444**, 51-59.
- 412 27. Biotage, *Extraction of Polyaromatic Hydrocarbons (PAHs) from Natural Waters Using  
413 ISOLUTE PAH Solid Phase Extraction Columns*, 2016.
- 414 28. J. W. Dolan, Calibration Curves , Part 1: To b or Not to b,  
415 [http://www.chromatographyonline.com/calibration-curves-part-i-b-or-not-  
416 b?id=&sk=&date=&pageID=5](http://www.chromatographyonline.com/calibration-curves-part-i-b-or-not-b?id=&sk=&date=&pageID=5), (accessed 15 May, 2015, 2009).
- 417 29. S. A. Tittlemier, T. Halldorson, G. A. Stern and G. T. Tomy, *Environmental Toxicology and  
418 Chemistry*, 2002, **21**, 1804-1810.
- 419 30. W. Laboratories, 2,2',3,3',4,5,5',6,6'-Nonabromo-4'-chlorodiphenyl ether (4PC-BDE-208),  
420 <http://www.well-labs.com/docs/pbde-analysis-4pc-bde-208.pdf>, 2009, (accessed 10 Jun  
421 2016).
- 422 31. USEPA, Method 525.3: Determination of semivolatile organic chemicals in drinking water by  
423 solid phase extraction and capillary column gas chromatography/mass spectrometry  
424 (GC/MS), [https://cfpub.epa.gov/si/si\\_public\\_record\\_report.cfm?dirEntryId=241188](https://cfpub.epa.gov/si/si_public_record_report.cfm?dirEntryId=241188), 2012,  
425 (accessed 12 Apr 2016).
- 426 32. J. Nácher-Mestre, R. Serrano, F. Hernández, L. Benedito-Palos and J. Pérez-Sánchez,  
427 *Analytica chimica acta*, 2010, **664**, 190-198.
- 428 33. USEPA, Guidelines Establishing Test Procedures for the Analysis of Pollutants, 40 CFR Part  
429 136, [https://www.gpo.gov/fdsys/granule/CFR-2016-title40-vol25/CFR-2016-title40-vol25-  
430 part136](https://www.gpo.gov/fdsys/granule/CFR-2016-title40-vol25/CFR-2016-title40-vol25-part136), 1995, (accessed 04 Nov 2015).
- 431 34. USEPA, Method 1614 brominated diphenyl ethers in water soil, sediment and tissue by  
432 HRGC/HRMS, [https://www.epa.gov/sites/production/files/2015-  
433 08/documents/method\\_1614a\\_2010.pdf](https://www.epa.gov/sites/production/files/2015-08/documents/method_1614a_2010.pdf), 2007, (accessed 01 May 2016).

- 434 35. X. Z. Peng, C. M. Tang, Y. Y. Yu, J. H. Tan, Q. X. Huang, J. P. Wu, S. J. Chen and B. X. Mai,  
435 *Environment International*, 2009, **35**, 303-309.
- 436 36. C. EU, *Off. J. Eur. Union*, 2013, **226**, 1-17.
- 437 37. B. O. Clarke, N. A. Porter, R. K. Symons, P. J. Marriott, G. J. Stevenson and J. R. Blackbeard,  
438 *Science of the Total Environment*, 2010, **408**, 1604-1611.
- 439 38. X. Wang, B. Xi, S. Huo, W. Sun, H. Pan, J. Zhang, Y. Ren and H. Liu, *Journal of Environmental*  
440 *Sciences*, 2013, **25**, 1281-1290.
- 441 39. M. Song, S. G. Chu, R. J. Letcher and R. Seth, *Environmental Science & Technology*, 2006, **40**,  
442 6241-6246.
- 443 40. K. D. North, *Environmental Science & Technology*, 2004, **38**, 4484-4488.
- 444 41. R. C. Hale, M. Alaei, J. B. Manchester-Neesvig, H. M. Stapleton and M. G. Ikonou, *Environment International*, 2003, **29**, 771-779.  
445
- 446 42. UNESCO, Managing water under uncertainty and risk,  
447 [http://www.unesco.org/new/fileadmin/MULTIMEDIA/HQ/SC/pdf/WWDR4%20Volume%201](http://www.unesco.org/new/fileadmin/MULTIMEDIA/HQ/SC/pdf/WWDR4%20Volume%201-2012-Managing%20Water%20under%20Uncertainty%20and%20Risk.pdf)  
448 [-Managing%20Water%20under%20Uncertainty%20and%20Risk.pdf](http://www.unesco.org/new/fileadmin/MULTIMEDIA/HQ/SC/pdf/WWDR4%20Volume%201-2012-Managing%20Water%20under%20Uncertainty%20and%20Risk.pdf), 2012, (accessed 13 Mar  
449 2017).
- 450

451 Table 1. Names, abbreviations, physical and chemical properties of PBDE congeners  
 452 investigated in this study.

<b>PBDE</b>	<b>Acronym</b>	<b>Molecular formula</b>	<b>Molecular weight (g/mol) <sup>b</sup></b>	<b>Solubility (g/L) at 25°C, pH 7 <sup>b</sup></b>	<b>Log <i>K</i><sub>ow</sub> <sup>a</sup></b>
<b>2,4,4'-TriBDE</b>	BDE 28	C <sub>12</sub> H <sub>7</sub> Br <sub>3</sub> O	406.90	7.7 x 10 <sup>-4</sup>	5.88
<b>2,2',4,4'-TetraBDE</b>	BDE 47	C <sub>12</sub> H <sub>6</sub> Br <sub>4</sub> O	485.79	2.5 x 10 <sup>-4</sup>	6.77
<b>2,2',4,4',5-PentaBDE</b>	BDE 99	C <sub>12</sub> H <sub>5</sub> Br <sub>5</sub> O	564.69	6.2 x 10 <sup>-5</sup>	7.66
<b>2,2',4,4',6-PentaBDE</b>	BDE 100	C <sub>12</sub> H <sub>5</sub> Br <sub>5</sub> O	564.69	7.3 x 10 <sup>-5</sup>	7.66
<b>2,2',4,4',5,5'-HexaBDE</b>	BDE 153	C <sub>12</sub> H <sub>4</sub> Br <sub>6</sub> O	643.58	1.6 x 10 <sup>-5</sup>	8.55
<b>2,2',4,4',5,6'-HexaBDE</b>	BDE 154	C <sub>12</sub> H <sub>4</sub> Br <sub>6</sub> O	643.58	1.9 x 10 <sup>-5</sup>	8.55
<b>2,2',3,4,4',5',6-HeptaBDE</b>	BDE 183	C <sub>12</sub> H <sub>3</sub> Br <sub>7</sub> O	722.48	5.6 x 10 <sup>-6</sup>	9.44
<b>DecaBDE</b>	BDE 209	C <sub>12</sub> Br <sub>10</sub> O	959.17	1.4 x 10 <sup>-6</sup>	12.11

453 Adopted from a <sup>17</sup>, b <sup>7</sup>

454 Table 2. Comparison between the retention times and monitoring ions of PBDE congeners of  
 455 interest using GC-MS and GC-ECD.

<b>Analyte</b>	<b>GC-ECD t<sub>R</sub> (min)</b>	<b>GC-MS t<sub>R</sub> (min)</b>	<b>EI-MS/SIM (m/z)</b>	
			<b>Ion 1</b>	<b>Ion 2</b>
<b>BDE 28</b>	5.26	5.35	246	246
<b>BDE 47</b>	6.69	6.83	326	324
<b>BDE 99</b>	7.76	7.92	405.5	403.5
<b>BDE 100</b>	8.13	8.30	405.5	403.5
<b>BDE 153</b>	8.98	9.17	483.5	481.5
<b>BDE 154</b>	9.47	9.66	483.5	481.5
<b>BDE 183</b>	10.67	10.87	563.5	561.5
<b>BDE 209</b>	16.08	16.76	811.5	809.5

456 t<sub>R</sub> = retention time

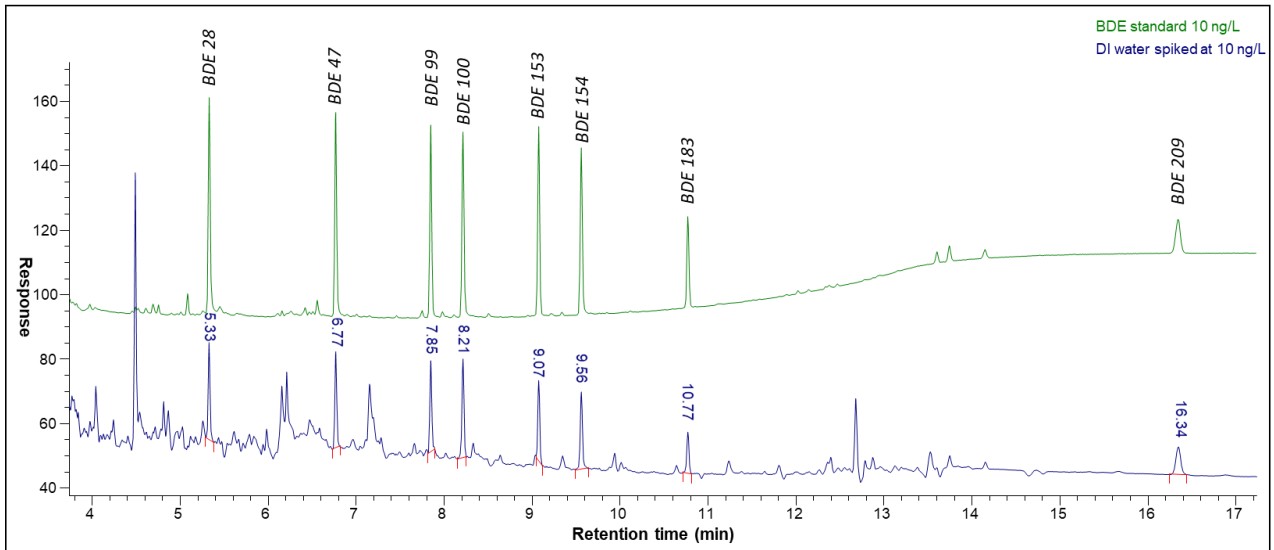
457

458

459

460

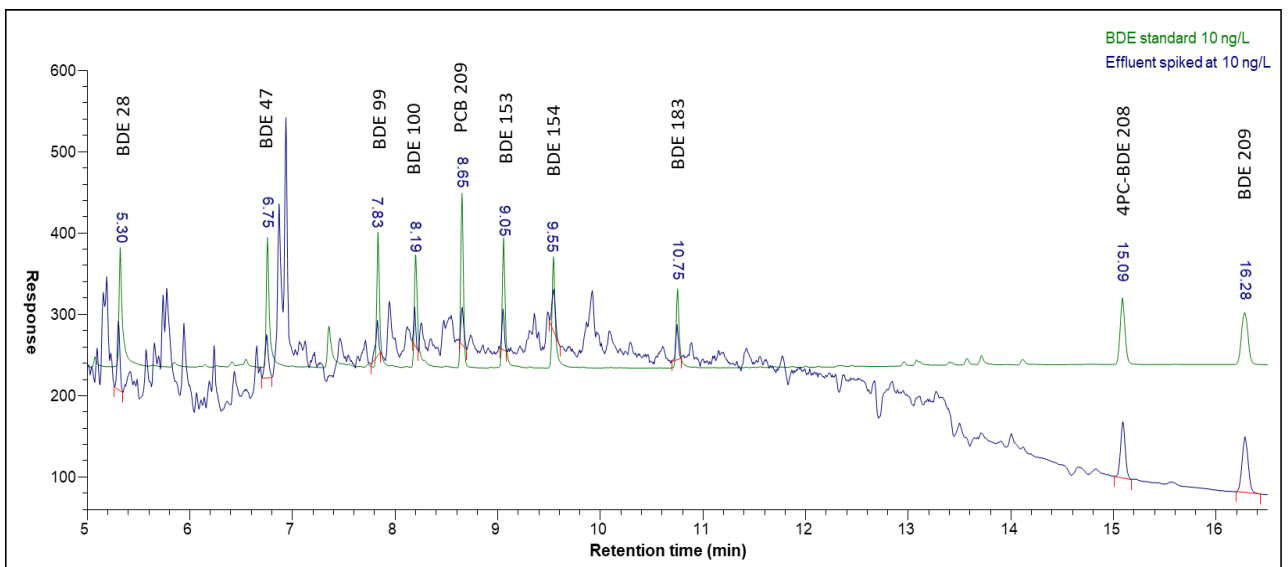
461



462

463 Figure 1. Example GC-ECD Chromatogram of eight primary BDE congeners spiked in  
464 deionized water (blue trace) at 10 ng/L (except BDE 209 spiked at 100 ng/L) over-layed by  
465 PBDE analytical standard (green trace). The PBDEs were spiked in deionized water before  
466 SPE.

467



468

469 Figure 2. Example GC-ECD Chromatogram of eight primary BDE congeners and two  
470 surrogate standards in effluent (blue trace) spiked at 10 ng/L (except BDE 209 and 4PC-  
471 BDE-209 spiked at 100 ng/L) over-layed by PBDE analytical standard (green trace). The  
472 PBDEs were spiked before SPE

473



474

475 Table 3. Determined calibration curve model and equations for the 8 PBDE congeners

Compound	R <sup>2</sup>	y-intercept	Standard Error (SE <sub>y</sub> )	Calibration model	Calibration equation
<b>BDE 28</b>	0.9992	0.106	0.049	y = mx + c	y = 0.1335x - 0.1065
<b>BDE 47</b>	0.9991	0.150	0.078	y = mx + c	y = 0.1391x - 0.1497
<b>BDE 99</b>	0.9980	0.150	0.083	y = mx + c	y = 0.1391x - 0.1501
<b>BDE 100</b>	0.9982	0.157	0.070	y = mx + c	y = 0.1242x - 0.1568
<b>BDE 153</b>	0.9993	0.010	0.045	y = mx + c	y = 0.0061x - 0.0398
<b>BDE 154</b>	0.9988	0.122	0.053	y = mx + c	y = 0.1152x - 0.1220
<b>BDE 183</b>	0.9978	0.085	0.042	y = mx + c	y = 0.0670x - 0.0847
<b>BDE 209</b>	0.9995	0.045	0.046	y = mx + c	y = 0.0058x - 0.0942

476

477

478

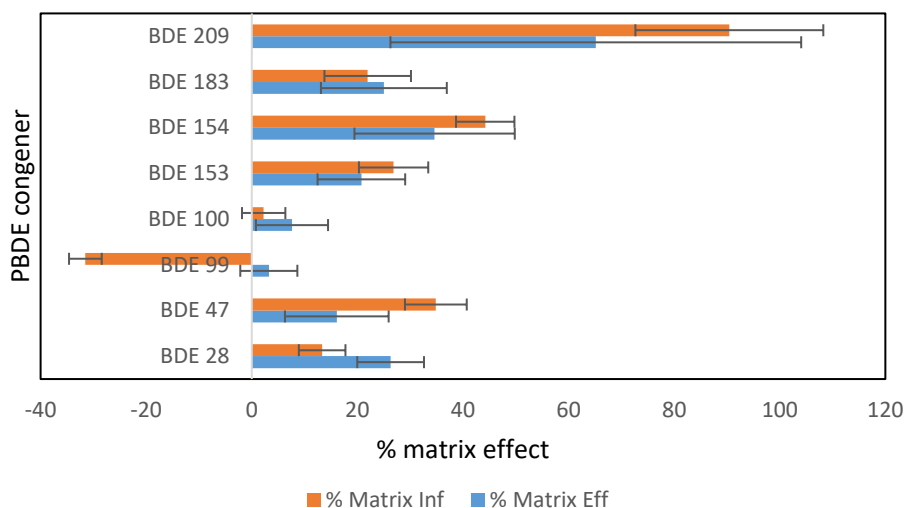
479

480 Table 4. Recoveries and relative standard deviation (RSD) of PBDEs spiked in deionised  
481 water and effluent at 10 ng/L. RSD in bracket

BDE	Recoveries in DI water (%)			Recoveries in effluent (%)		
	Isolute C 18	Isolute PAH	Oasis HLB	Isolute C 18	Isolute PAH	Oasis HLB
<b>BDE 28</b>	85 (1.7)	54 (0.7)	60 (7.1)	71 (12.5)	78 (5.8)	46 (8.1)
<b>BDE 47</b>	75 (8.2)	61 (13.9)	65 (7.5)	84 (10.9)	79 (0.6)	44 (13.1)
<b>BDE 99</b>	76 (16.7)	73 (1.6)	70 (9.1)	63 (2.1)	82 (0.8)	52 (10.6)
<b>BDE 100</b>	71 (17.8)	58 (3.3)	79 (4.5)	52 (1.0)	69 (0.2)	33 (0.6)
<b>BDE 153</b>	59 (18.1)	56 (0.9)	53 (10.1)	45 (2.5)	94 (8.9)	42 (1.8)
<b>BDE 154</b>	58 (19.6)	53 (1.2)	51 (6.3)	70 (27.2)	97 (12.5)	59 (8.6)
<b>BDE 183</b>	59 (8.3)	60 (2.7)	62 (7.9)	53 (12.7)	126 (0.1)	35 (3.5)
<b>BDE 209</b>	52 (23.8)	96 (0.1)	65 (10.2)	169 (13.3)	90 (12.8)	122 (6.2)
<b>AVERAGE</b>	66.88	63.88	63.13	75.88	89.38	54.13
<b>SD</b>	11.46	14.40	9.00	39.66	17.44	28.69

482

483



484

485 Figure 3. Observed matrix effect in the analysis of PBDEs in influent and effluent samples. Inf  
 486 and Eff represents influent and effluent respectively

487

488

489 Table 5. Surrogate corrected recovery at 3 ng/L and 10 ng/L in effluent, instrumental  
 490 detection limit (IDL) and method detection limit of PBDE congeners in deionized (DI) water  
 491 and effluent. Surrogate standards are indicated in italics.

Chemical	Corrected recovery in effluent (%)		IDL* (pg/μl)	MDL (ng/L)	
	3 ng/L	10 ng/L		DI water	Effluent
<b>BDE 28</b>	78 (16.1)	90 (7.0)	0.2	0.68	0.66
<b>BDE 47</b>	176 (11.7)	129 (15.2)	0.2	0.54	2.57
<b>BDE 99</b>	74 (6.4)	116 (4.7)	0.5	0.33	2.54
<b>BDE 100</b>	62 (5.1)	83 (7.3)	0.5	0.27	1.89
<b>BDE 153</b>	78 (1.8)	112 (9.1)	0.5	0.14	0.20
<b>BDE 154</b>	131 (7.5)	60 (6.7)	0.5	0.44	4.19
<b>BDE 183</b>	86 (8.2)	150 (6.7)	1.0	0.53	1.31
<b>BDE 209</b>	135 (3.4)	113 (2.3)	5.0	10.04	10.76
<i>PCB 209</i>	-	-	0.2	0.28	1.09
<i>4PC-BDE-208</i>	-	-	5.0	4.37	6.12

492 \*IDL was determined with PBDE analytical standard

493

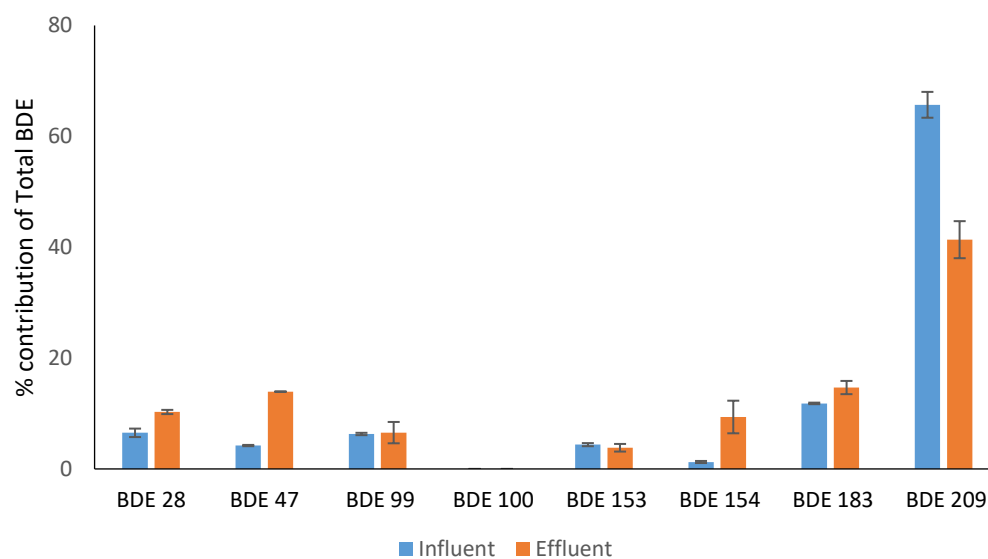
494

495 Table 6. Concentration of BDE congeners in effluent and influent of a UK activated sludge  
 496 WWTP

<b>PBDE congener</b>	<b>Influent (ng/L)</b>	<b>Effluent (ng/L)</b>	<b>% Removal</b>
<b>BDE 28<sup>¶</sup></b>	11.0 (1.3)	4.4 (0.2)	60.0
<b>BDE 47<sup>¶</sup></b>	7.1 (0.2)	6.0 (0.1)	15.5
<b>BDE 99<sup>¶</sup></b>	10.7 (0.4)	2.8 (0.8)	73.8
<b>BDE 100<sup>¶</sup></b>	< 1.89	< 1.89	-
<b>BDE 153<sup>¶</sup></b>	7.1 (0.6)	1.6 (0.3)	77.5
<b>BDE 154<sup>¶</sup></b>	2.1 (0.3)	4.0 (1.3)	0
<b>BDE 183</b>	20.0 (0.2)	6.3 (0.5)	68.5
<b>BDE 209</b>	111.3 (10.7)	17.7 (1.4)	84.1
<b>∑PBDE</b>	169.4	42.9	74.7
<b>∑PBDE EU MAC-EQS</b>	-	18.8	29.9*

497 <sup>¶</sup> PBDE congeners included in the EU WFD \* Average % removal

498



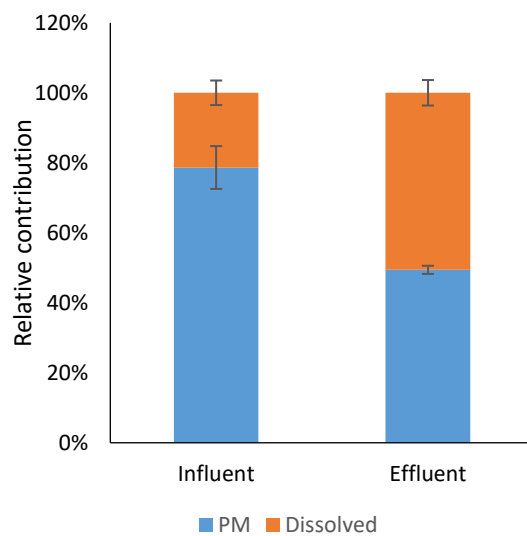
499

500 Figure 4. Distribution of BDE congeners in influent and effluent samples from a UK WWTP

501

502

503



504

505

506

507

Figure 5. Partitioning of PBDEs over particulate matter (PM) and dissolved phases of influent and effluent samples from a UK wastewater treatment plant.

508